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## PHOTOCHEMICAL ALLYLIC

## AND DIALLYLIC REARRANGEMENTS

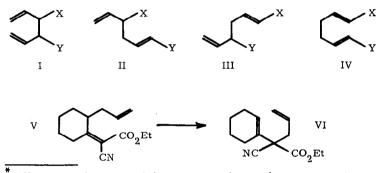
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We have discovered a range of photochemical allylic rearrangements, which we would like to illustrate with some typical examples.\*

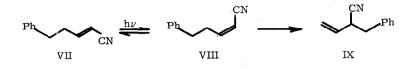
In derivatives of diallyl (e.g. I); either one (II and III) or both allyl groups (IV) may shift. Irradiation of  $V^1$  in cyclohexane with unfiltered light from a medium-pressure mercury arc caused rearrangement in good yield into the unconjugated isomer VI.



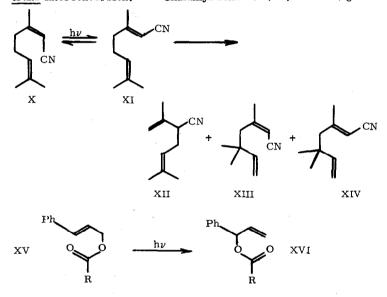
\* All compounds gave satisfactory u.v., i.r. and p.m.r. spectra, and the structures were confirmed by unambiguous syntheses.

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Under the same treatment the nitrile VII was converted into the <u>cis</u>isomer VIII, and, more slowly, into the isomer IX in which the benzyl group had migrated (23% yield).



Geranchitrile<sup>2</sup> (X) rearranged into the isomers XII, XIII and XIV, but, as expected, the 6,7-dihydro-derivative was unchanged beyond <u>cis</u>trans interconversion. Cinnamyl benzoate (XV; R = Ph) gave



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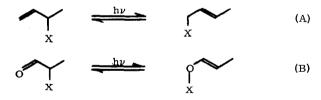
 $\alpha$ -phenylallyl benzoate (XVI) in good yield apart from some polymerisation. The much slower rearrangement of cinnamyl acetate (XV; R = Me) to (XVI; R = Me) suggests that the benzoyl group plays a significant part in absorbing the light.

Thermally these reactions go in the reverse direction (e.g.  $VI \longrightarrow V$ ;  $XVI \longrightarrow XV$ ), so that in some cases the photochemical transformation may be useful in making the less stable, unconjugated isomer.

Although, naturally enough, cinnamyl acetate in cyclohexane was not affected by light filtered through Pyrex glass, in acetone rearrangement to (XVI; R = Me) proceeded. An even better yield (65%) of the benzoate was obtained with filtered light by using benzene containing cinchonine. These photo-sensitised reactions presumably take place through the triplet state of the styrene group<sup>3</sup> followed by cleavage into a pair of radicals that recombine, probably without diffusing apart. In the related photo-equilibrium<sup>4</sup> between  $5\alpha$ -acetoxy- $\Delta^3$ - and  $3\alpha$ -acetoxy- $\Delta^4$ -cholestan-7-one there is a ketone group within the molecule, which may have the same function. Whether the two oxygen atoms become equivalent during the migration will be settled only by their isotopic identification, but a concerted shift through a six-membered cyclic transition-state would be contrary to the symmetry of the orbitals

in the first excited singlet state<sup>5</sup> (although the geometrically more difficult shift through a four-membered transition-state without inversion of the carboxy group would be allowed).

Although nothing is yet known about the mechanisms of these reactions, it is possible that many of them too proceed through the triplet state and diradicals.



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